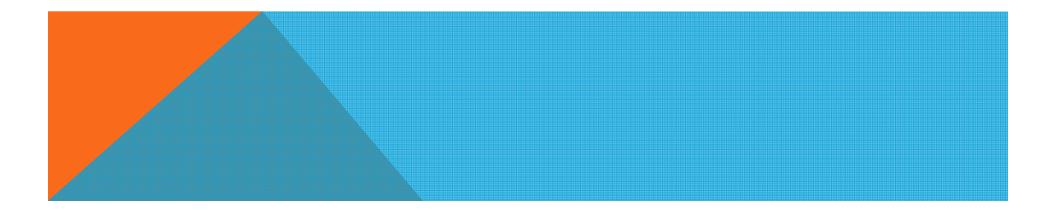
TIO₂ PHOTOCATALYSIS FOR WATER PURIFICATION

ELEANOR SOLOMON SIMON FOWLER DR. JUN JIAO

PURPOSE

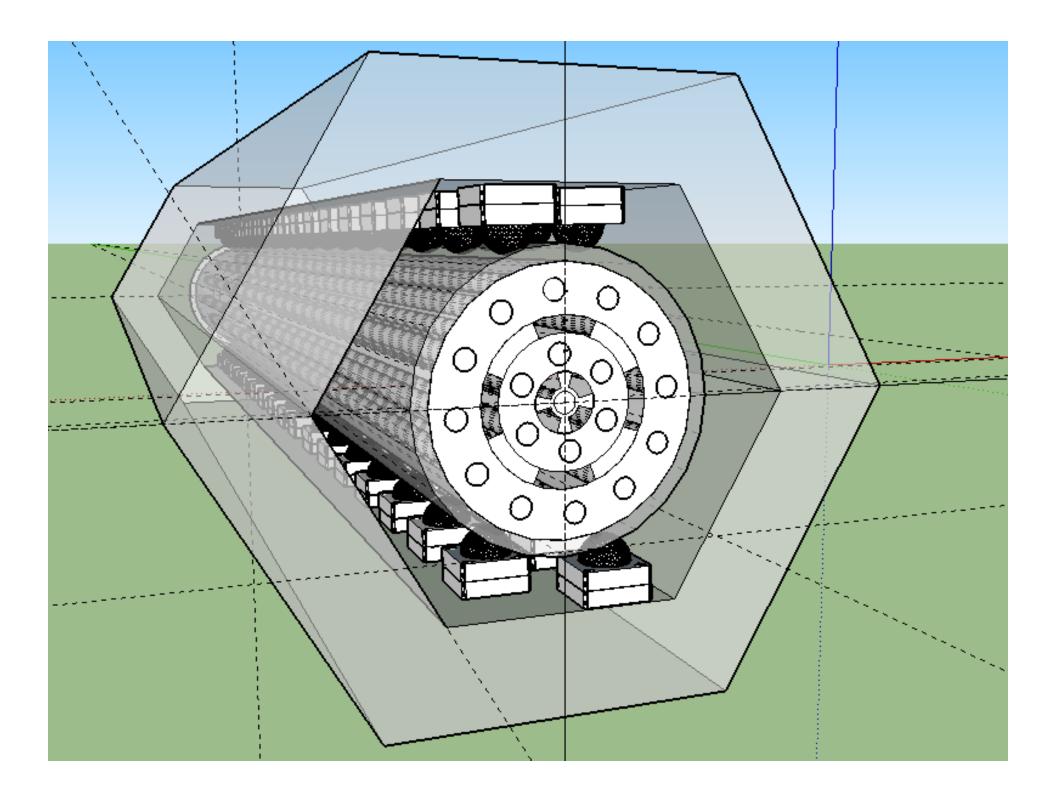
- Build a system that purifies water using light
- Save water!



DEFINITIONS

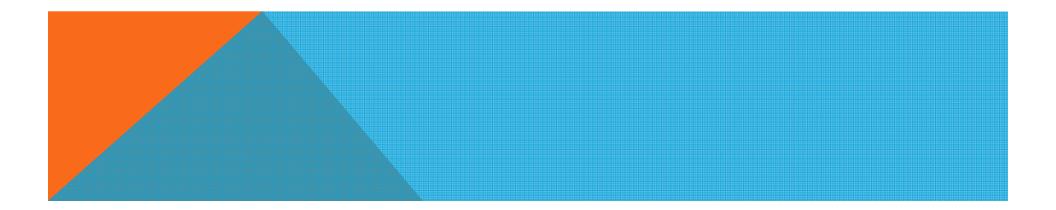
- Photocatalysis A reaction caused by the interaction between light and the catalyst.
- Purification Breaking down all organic contaminants into simple molecules.
- Titanium Dioxide Our catalyst.





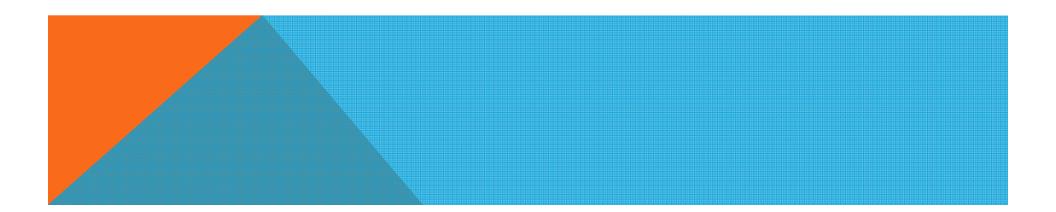
THE PROCESS

- Light shines onto TiO₂
- A photon from the light excites an electron from the TiO_2 and leaves an electron hole.



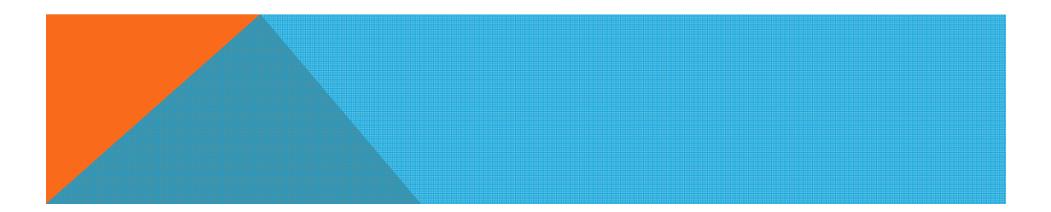
ELECTRON HOLES

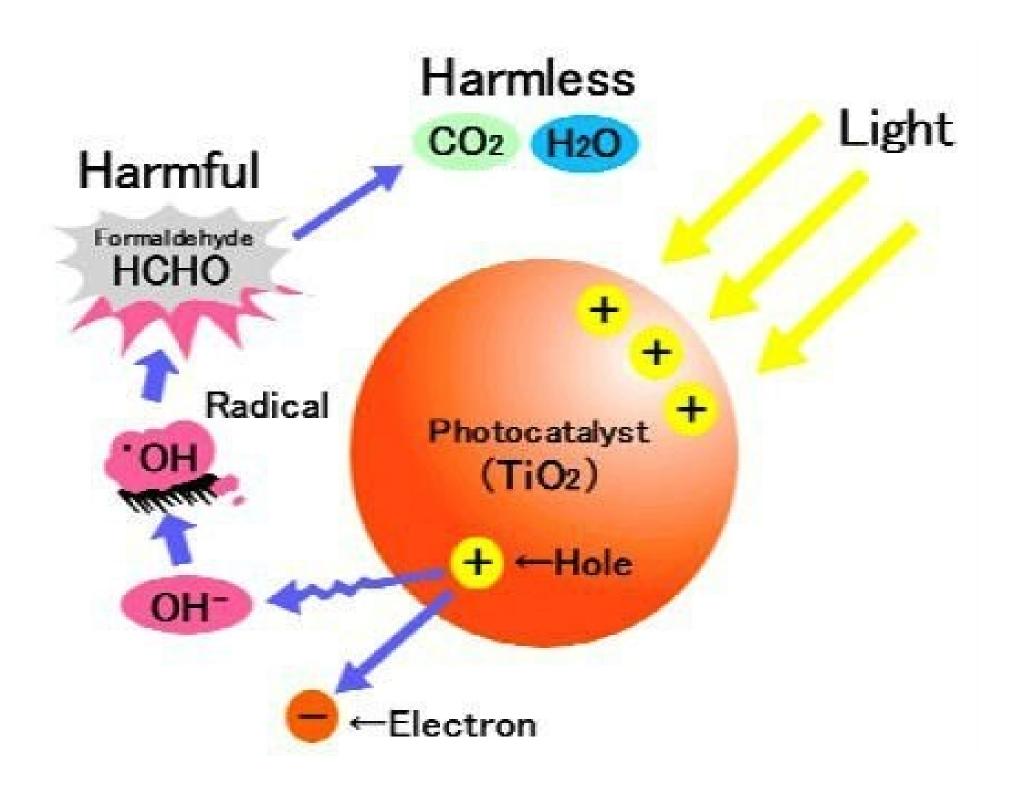
- Electron hole a space that an electron leaves behind when it is taken away from a valence band.
- Acts as a particle because of the way other particles react with it.
- Has a positive charge.



THE PROCESS CONTINUED

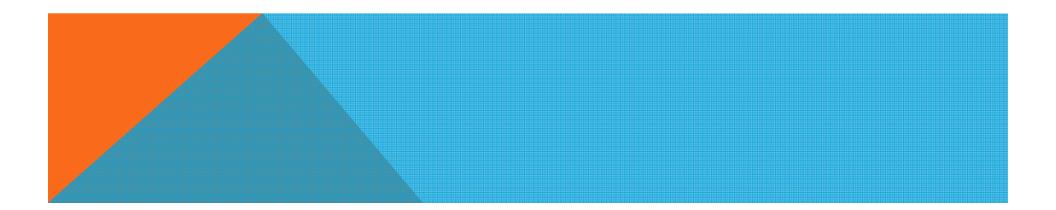
- The electrons and electron holes react with the water and create two things:
- Hydroxyl radicals OH
- Superoxide anions O_2^-
- Hydroxyl radicals and superoxide anions react with contaminants.
- Break down contaminants.





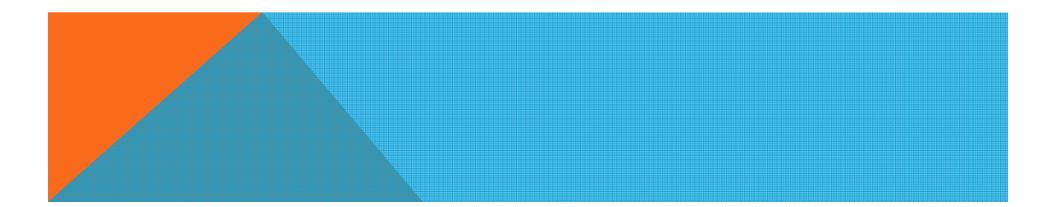
COATING

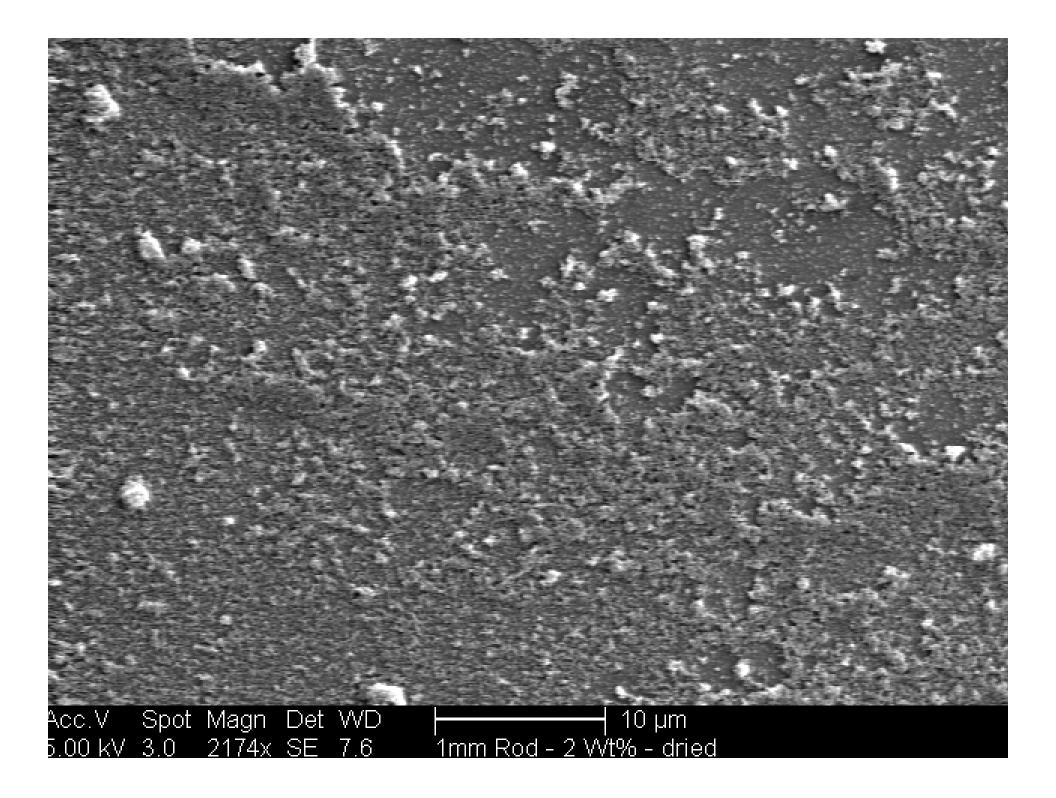
- Coated quartz rods in TiO₂ mixture.
- TiO₂ mixture had different weight percentages of TiO₂
- Question: What weight percentage of TiO₂ works best to coat the quartz rods?
- Tested: 2%, 4%, 6%, and 8% of TiO₂

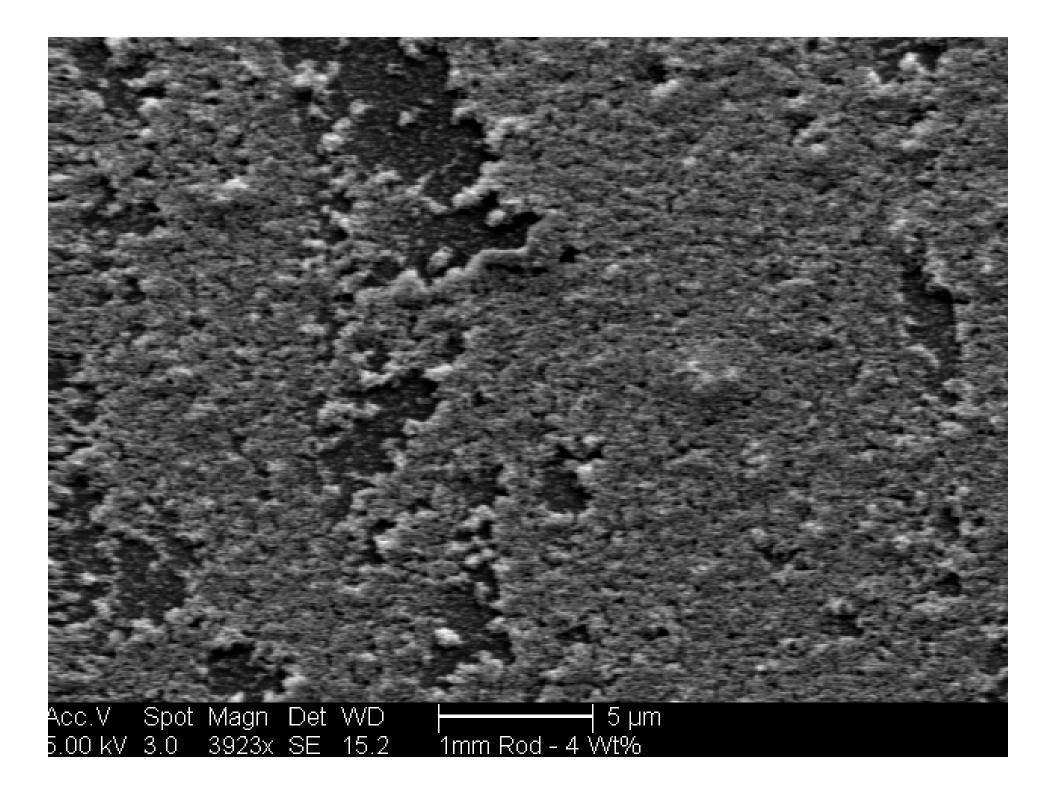


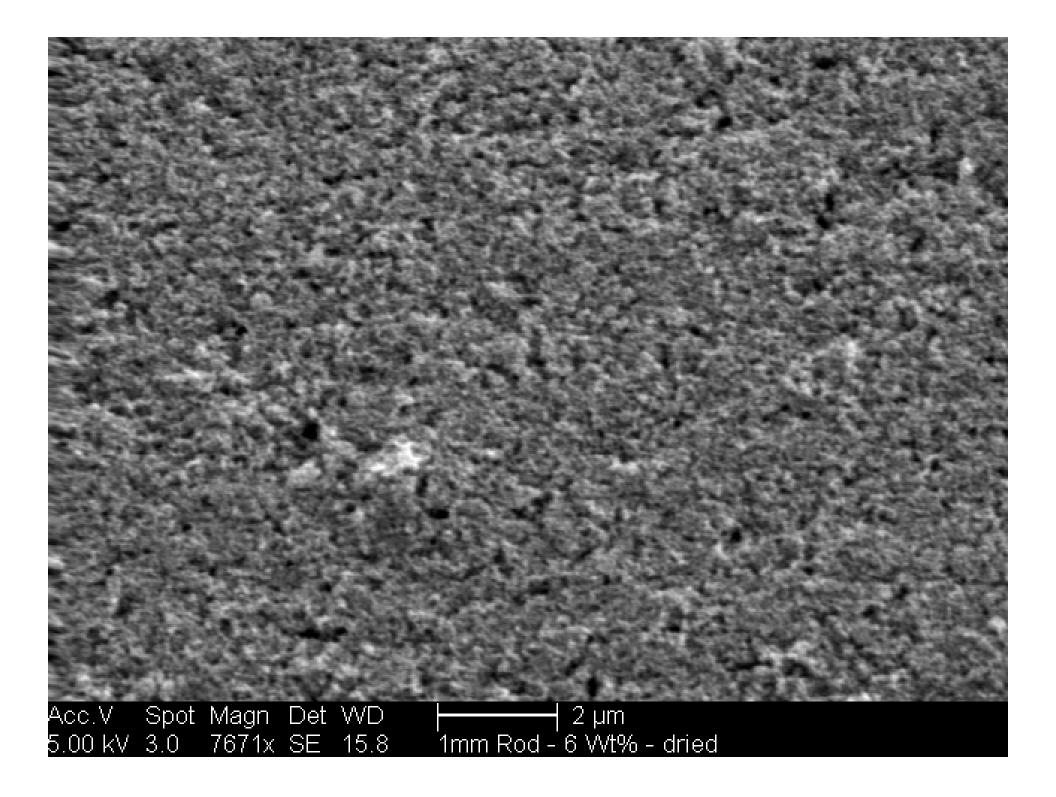
PROCEDURE

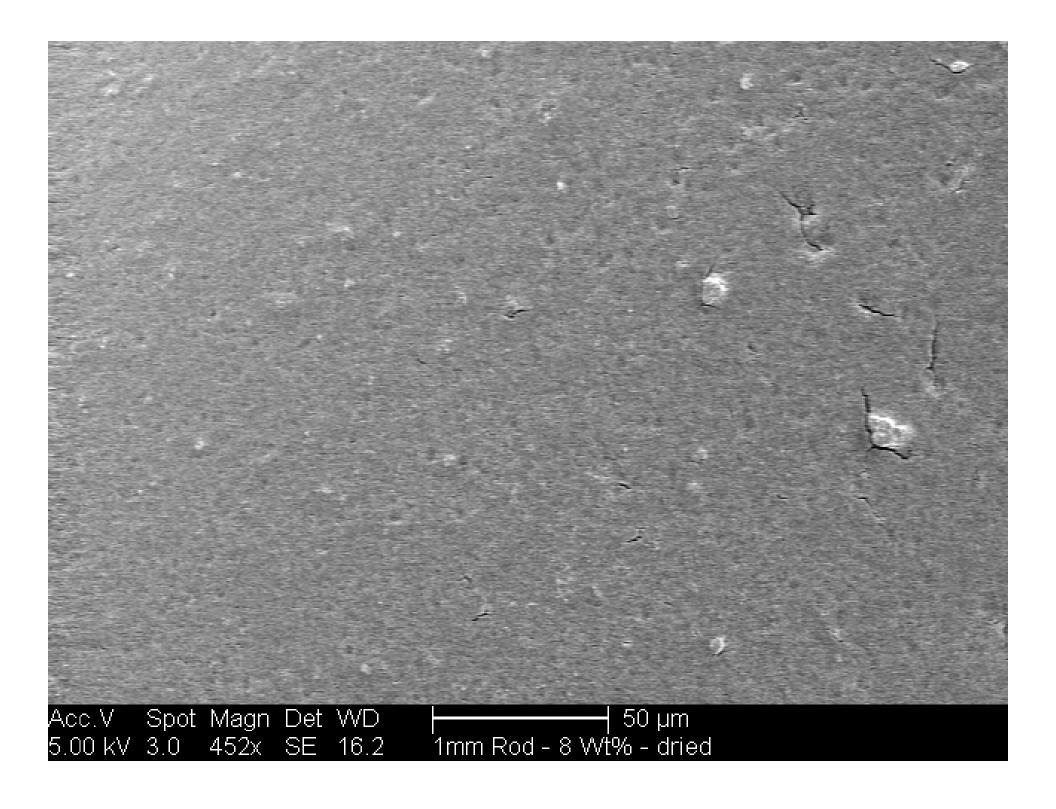
- Broke 1 mm rod into pieces each 35 mm long
- Coated in TiO₂ mixtures
- Baked in oven for 4 hours at 500 degrees C.
- Took out and looked at under the SEM





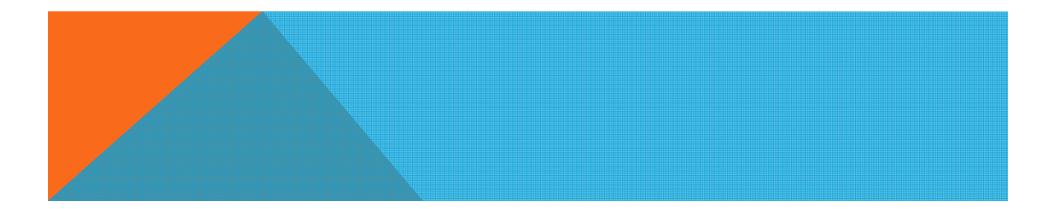






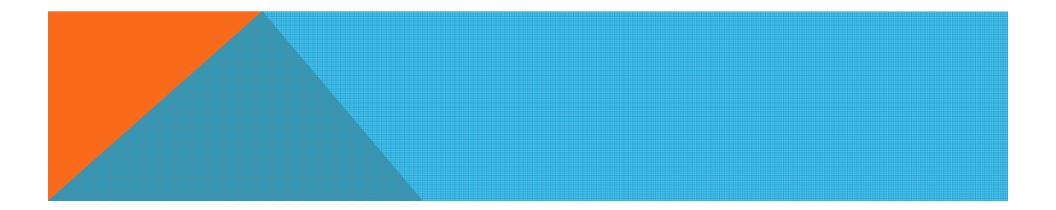
FINDINGS

- 2% and 4%: Spotty coating
- 6%: Mostly smooth and uniform
- 8%: Showed signs of cracking from baking



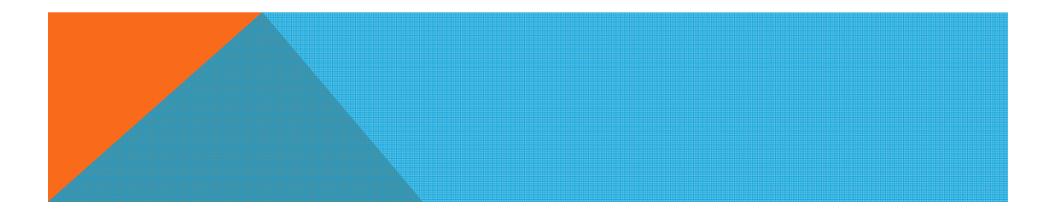
CONCLUSIONS

- 6% is the best option
- 4% could work
- 8% also could work if we use different drying procedure instead of baking



GOALS FOR THE REST OF THE SUMMER

- Use a dip coater to coat the rods
- Assemble a full reactor



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