

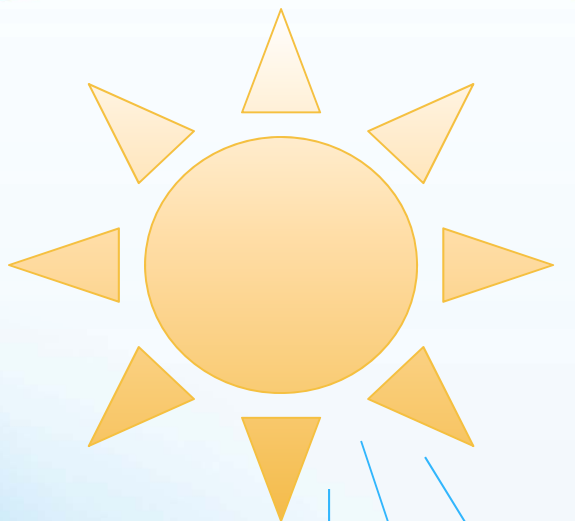
TiO₂ as a Catalyst in Thin Film Coatings of Glass Beads for Water Purification Purposes

- * *Presented By: Sydney Quinton-Cox*
- * *Bioengineering Department at Oregon State University*
- * *Research Experience for Undergraduates~2014*
- * *Dr. Jiao's Lab at PSU*

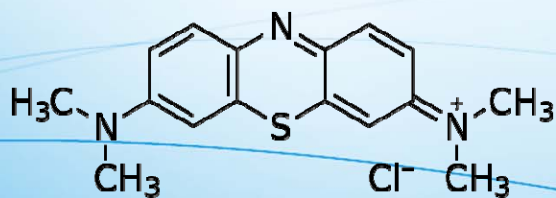
Overview

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Background

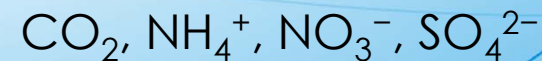


- * The sun breaks down organic compounds using UV light.
- * TiO_2 is a catalyst for this reaction.



Methylene Blue Dye

The Methylene Blue Dye is broken up into safe compounds.



Why this Photocatalyst is Needed

Pharmaceuticals

In 2011

- * *4,020,000,000 prescriptions were written.*

In 2013 a study found

- * *70% take at least 1*
- * *50% have at least 2*
- * *20% are on 5 or more*
- * *Over the counter drugs*

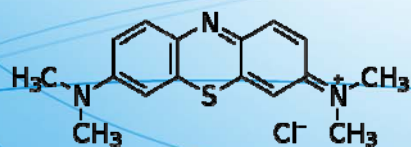


Cosmetics



Pesticides & Herbicides

Dyes



Methylene Blue Dye



Current Techniques in Industry

* Bacteria

- Contained in large pools
- Large quantities of oxygen
- A slow process



* Filtering Systems

- Difficult to do
- Leaves behind concentrated waste

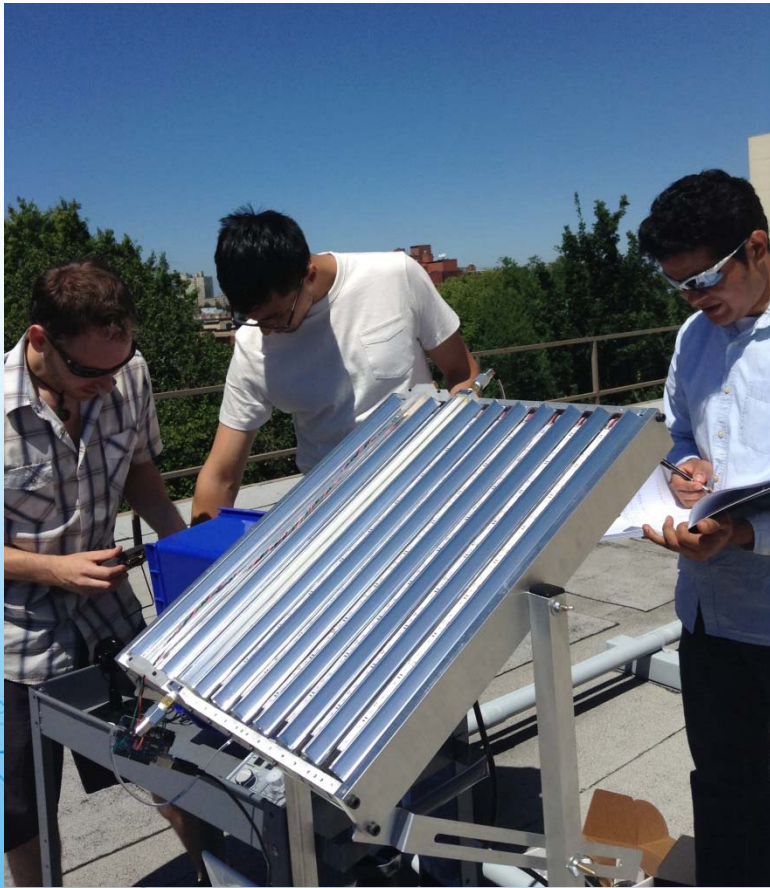


* UV Producing Light Bulbs

- Large pools of water
- High energy costs



Research Group



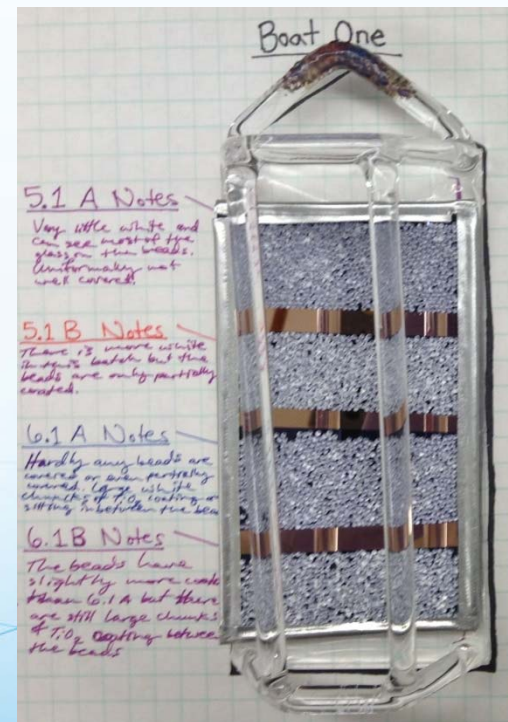
- * The Team

- Simon/ Physics
- Emilio/ Mechanical Engineering
- Esteban/ Chemistry
- Sydney (Me)/ Bioengineering

- * The project that we are all working on is the prototype to the left.

My Research

- * Borosilicate Glass Beads (1mm)
 - performed synthesis trials
 - dried the beads
 - sintered the beads
 - assessed their appearance
 - SEM work on Tuesday nights
 - degradation trials
 - Raman tests



Drying Machines

- * Before baking the beads they must be dried at a low temperature.
- * This decreases cracking or flaking off of the TiO_2 coating.



- * Drying Techniques

- air drying
- ceramic oven (40°C)
- tumble drying

- * Tumble Dryer

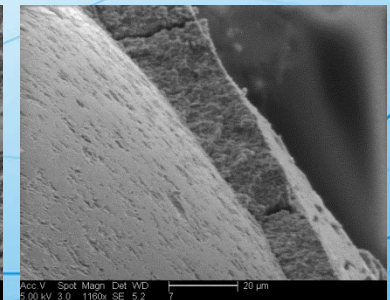
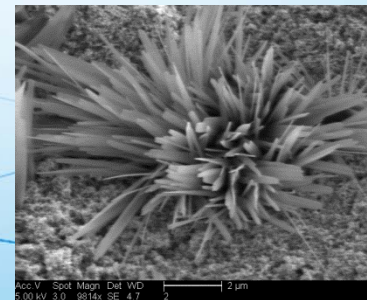
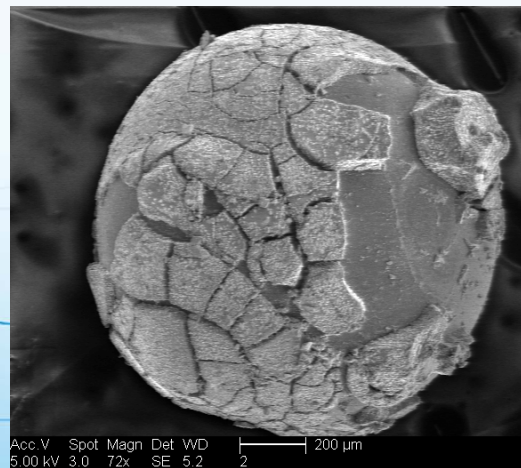
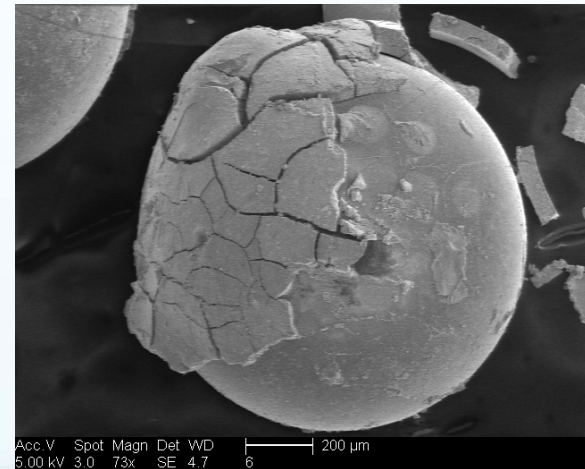
- repurposed a rock polisher
- designed insert on AutoCAD (w/ Emilio)
- bought marbles to move the beads



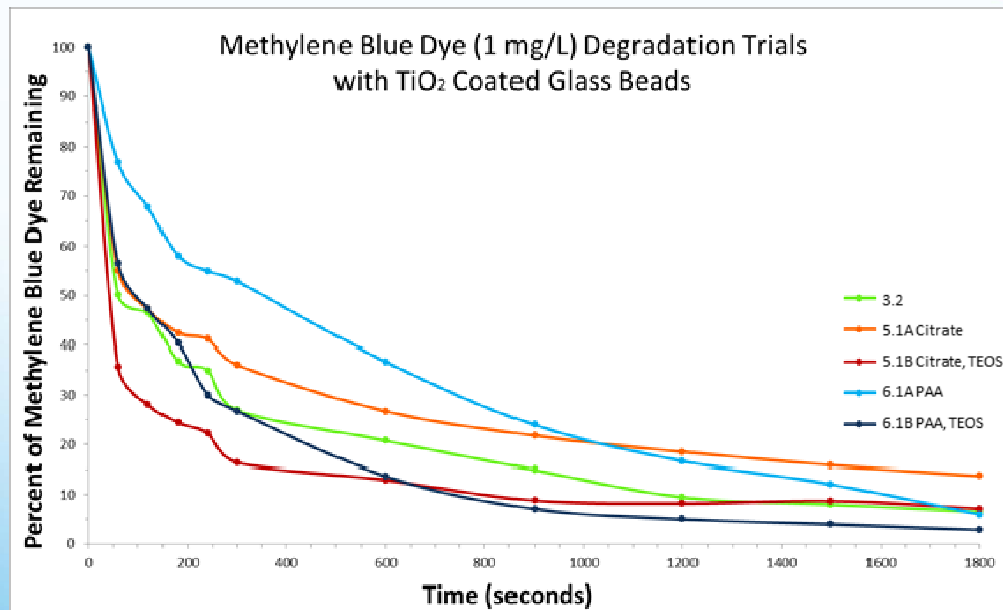
Scanning Electron Microscope



- * SEM Tuesday Nights
- * Surface Morphology
 - cracking
 - thickness
 - contaminants
- * Detector for Spectra

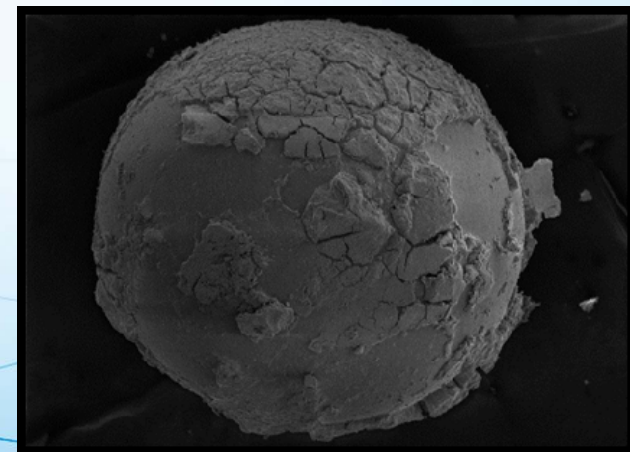


My Final Project



Ingredients	Original Synthesis	Trial 1	Trial 2	Trial 3
TiO ₂ (g)	4.0162	1.5010	1.4993	1.501
Water(mL)	7	11	15	13
TEOS(mL)	1.0	1.0	1.5	1.0
PAA(μL)	220	200	200	300
Glass Beads(g)	7.0063	20.095	20.006	20.003

SEM photo of a bead from the original PAA & TEOS synthesis

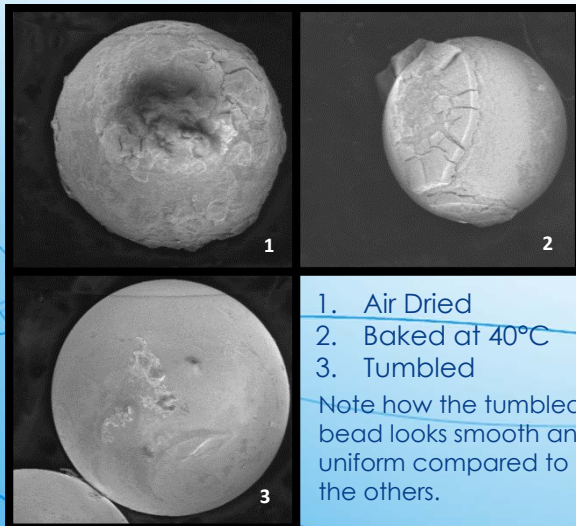


Drying Process & Sintering Results

Drying

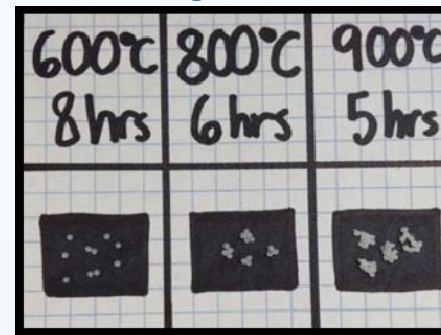
Drying Technique	Notes
Air Dried	-Visible cracking -Large rings formed where the bead sat
Baked at 40°C	-Dried faster -Visible cracking -Large rings formed where the bead sat
Tumbled w/ 14 Marbles	-Smooth surface -Thinner Coating -Takes longer to drying if not vented

Method of Choice: Tumbling

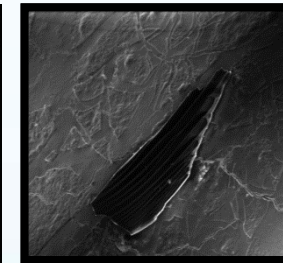


1. Air Dried
 2. Baked at 40°C
 3. Tumbled
- Note how the tumbled bead looks smooth and uniform compared to the others.

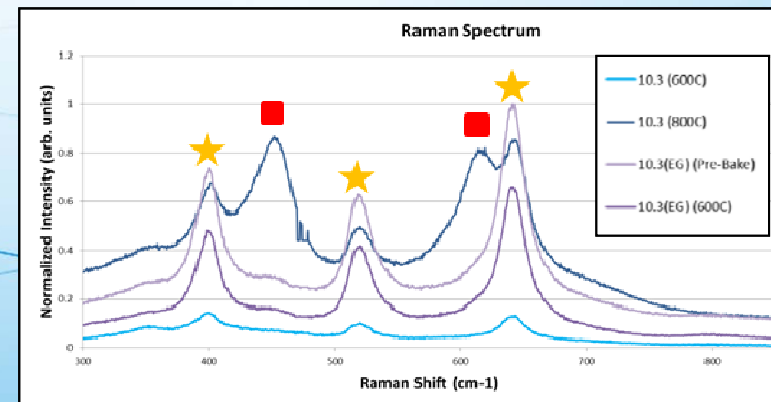
Sintering Temperatures



Trial 2 baked at each sintering temperature.



Piece of silicon wafer adhered to the TiO₂ coating on a bead after being baked at 900°C.



The graph was normalized at the 635 1/cm peak.

Best Bead Coating

Process of Choice

Synthesis:

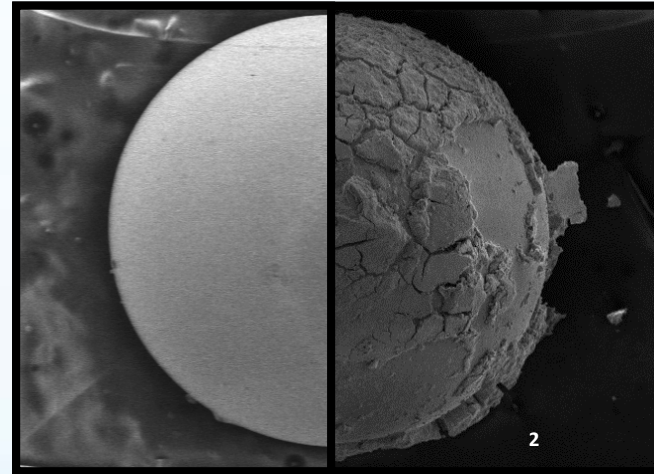
- * Trial 3 (has more PAA)

Drying Technique:

- * Tumbling

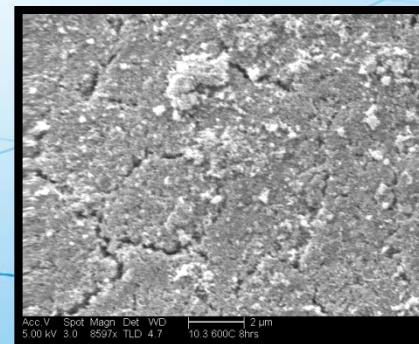
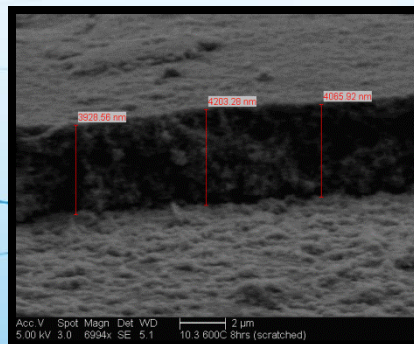
Sintering Process:

- * 600°C for 8 Hours



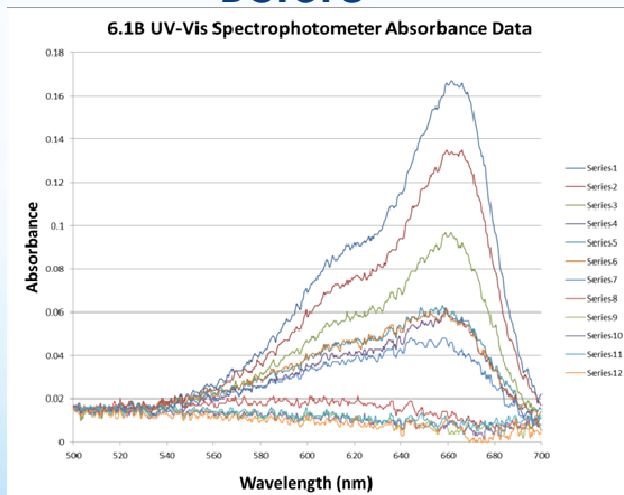
Created

- * 4 micron thin coating
- * Small nanometer cracks
- * No flaking



Degradation Results

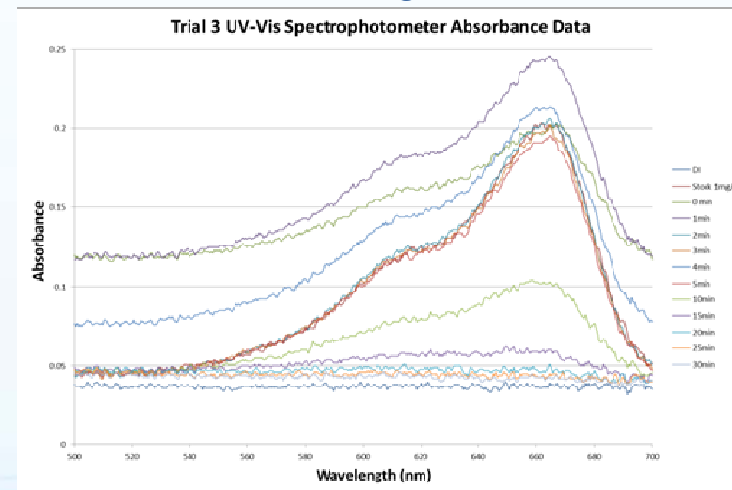
Before



The 6.1B degradation was done again at the same conditions as Trial 3.

- * Light intensity AM 1.5
- * 1 milligram of methylene blue dye per liter
- * 16mL of contaminated water per degradation trial

After



The stir bar was turned on after the 5 minute sample. Top curves staying constant confirmed our assumption that circulation is important.

- * more accurate reading
- * brings the contaminants to the beads where they are degraded

Less TiO₂ in the water

Potentially a faster degradation

Goal of The Group Project

- * Create a more efficient way to clean water.
- * A less expensive and more eco friendly solution.
- * To create a product that is commercially viable.



Acknowledgments

- * Pete & Rosalie Johnson
- * Dr. Skip Rochefort
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- * Simon Fowler
- * Emilio & Esteban
- * Lester Lampert
- * Meaghan Corwin



References

* Information:

- * <http://www.sciencedirect.com/science/article/pii/S0926337300002769>
- * <http://www.cbsnews.com/news/study-shows-70-percent-of-americans-take-prescription-drugs/>
- * <http://www.usatoday.com/story/money/business/2013/07/28/deadly-epidemic-prescription-drug-overdoses/2584117/>
- * http://apps.who.int/iris/bitstream/10665/44630/1/9789241502085_eng.pdf?ua=1

* Photo Credit:

- * <http://res.mindbodygreen.com/img/ftr/glass-of-water.jpg>
- * <http://www.nsf.gov/images/logos/nsf1.gif>
- * http://upload.wikimedia.org/wikipedia/commons/thumb/9/9f/Methylene_blue.svg/640px-Methylene_blue.svg.png
- * <http://gascylinders.files.wordpress.com/2012/05/sasasas1.jpg>
- * <https://store.thinksai.com/images/clipart/Safety%20First/Danger/Horizontal/dangH134%20-%20hazardous%20waste%20handle%20with%20care.jpg>
- * <http://www.dfwais.com/wp-content/uploads/2013/04/energy-savings.jpg>
- * <http://sleepdisorders.dolyan.com/wp-content/uploads/2012/05/Do-Birth-Control-Pills-Cause-Sleep-Disorders.jpg>
- * <http://rudoilaw.com/wp-content/uploads/2013/06/scripts.jpg>
- * <http://blog.heritage-enviro.com/Portals/31121/images/herbicide%20spraying.jpg>
- * <http://images.justbathroomsigns.com/img/lg/K/Warning-Pesticides-Sign-K-2922.gif>
- * http://freerb.ru/images/2013/06/26/108861/dekorativnaya-kosmetika_1.jpg