

The Optimization of TiO₂ Thin Films for Photocatalytic Water Purification

Anthea Leng
The Jiao Lab
Portland State University

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Build an efficient water purification system through photocatalysis using UV light and titanium dioxide.

Decreasing the amount of pollutants in water supplies and having safer drinking and usable water.

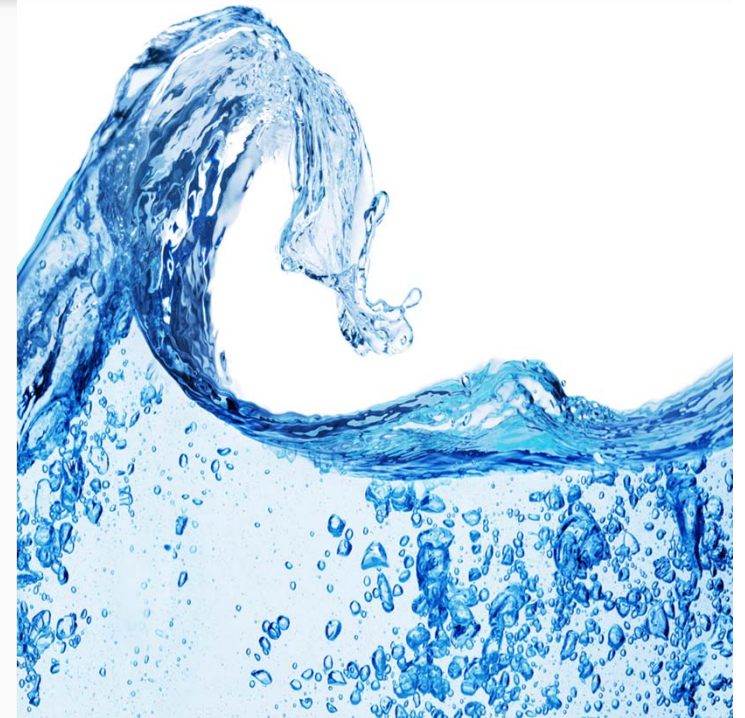


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783 million people do not have access to clean water. [1]

6-8 million people die a year from water related diseases and disasters.[2]

Many water purification systems are costly and not efficient.



Photocatalysis

Catalyst: A substance that speeds up a chemical reaction

Photocatalysis: A catalytic process facilitated by light



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Xli\$erh\$ket

Two energy bands in TiO₂

Valence band (V_b)- holds outer electrons

Conduction band (C_b)- The band that accepts the electrons from V_b . When electrons move to C_b , they become mobile and transport charge^[3, 4]

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Electron can be raised from V_b to C_b [3, 4]

Has become excited

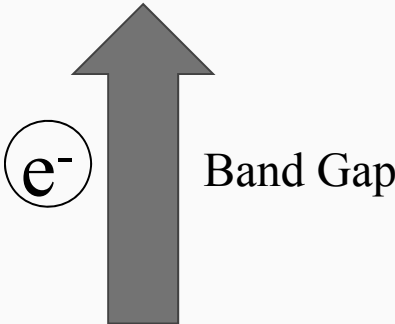
C_b will have a negative charge due to gaining of electron

V_b will have a positive charge in the form of an electron hole

Electron holes facilitate the oxidation of a pollutant

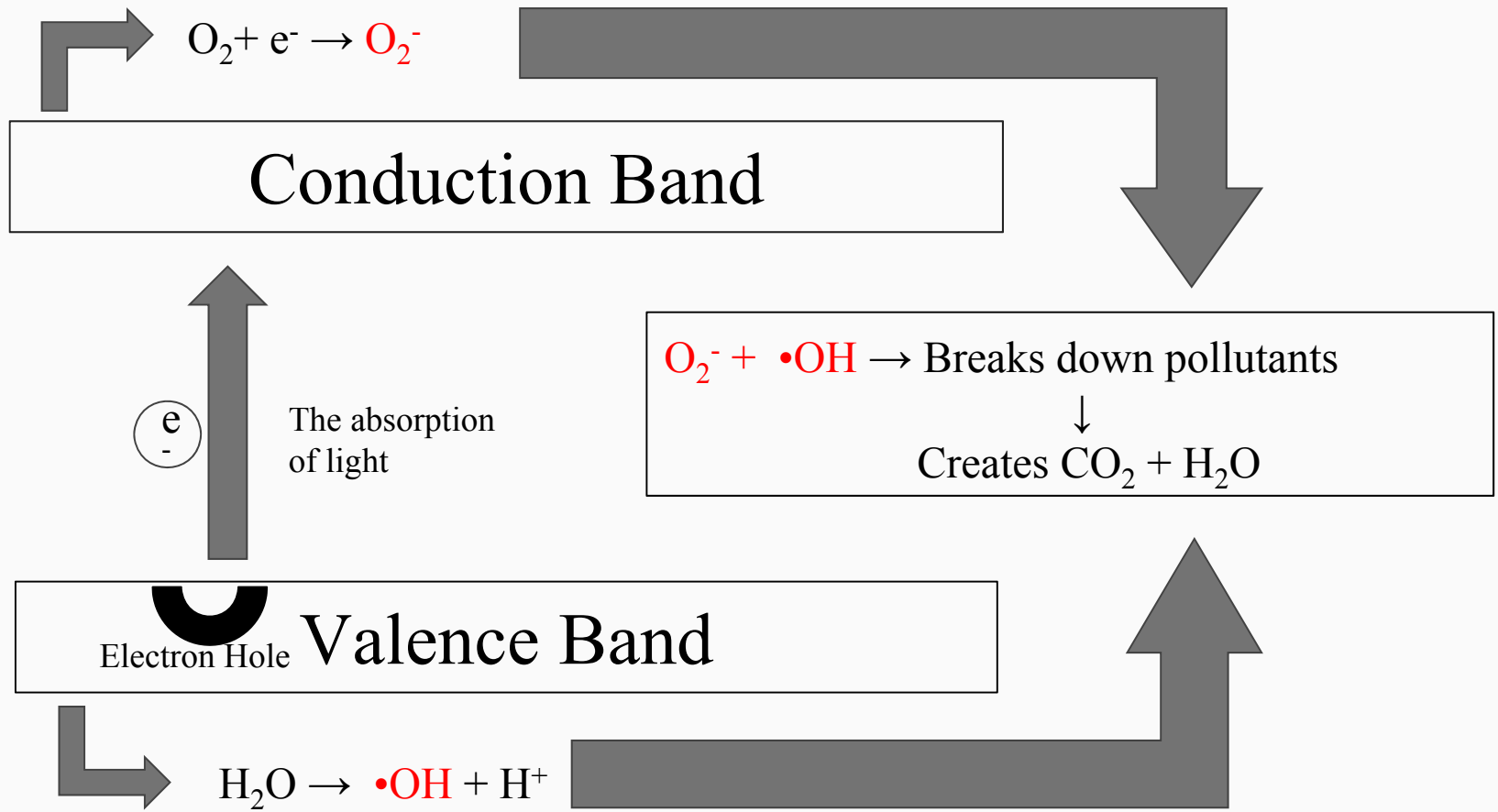
Oxidation purifies the water

Conduction Band



Electron hole

Valence Band



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A semiconductor

Harmless to the environment and
human/animal population^[3]

Cost effective and easily attainable^[3]



Synthesis

- Sol-Gel coating precursors
 - Titanium Tetraisopropoxide (TTIP)
 - Acetylacetonate (ACAC)
 - Ethanol Alcohol (EtOH)
- The catalyst
 - Titanium dioxide (TiO_2)- formed after calcination



The sol-gel before TiO_2 is added

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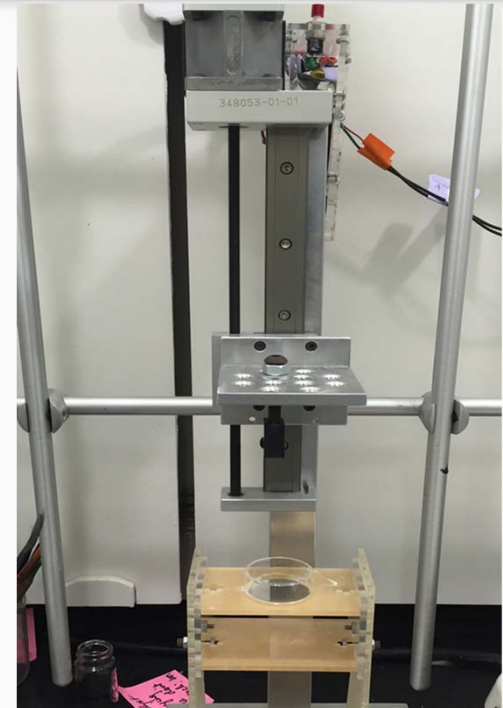
Magnetically stir the TTIP, ACAC, EtOH together to thoroughly combine each precursor

Add TiO_2 particles to the sol-gel

Clean quartz slide substrates

Dip each slide in the sol-gel solution

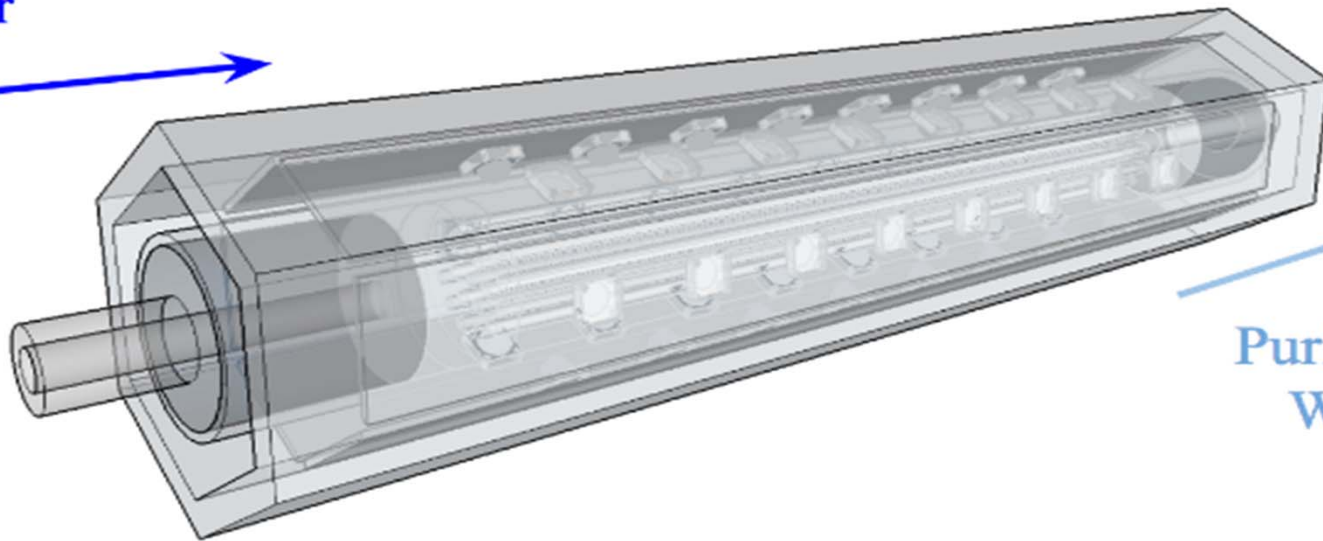
Dry slides



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Build a reactor with dip-coated rods and UV-lights for photocatalytic reaction to take place in and degrade the contaminants inside the water.^[5]

Contaminated
Water



Purified
Water

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How does the thickness of the
sol-gel coating affect
photocatalytic ability?

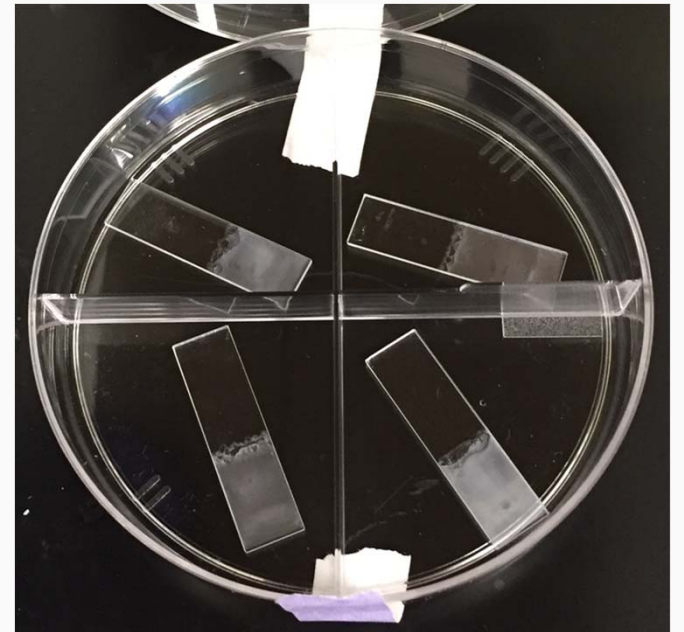
X-ray photoelectron spectroscopy

Manipulated variable: Coating thickness

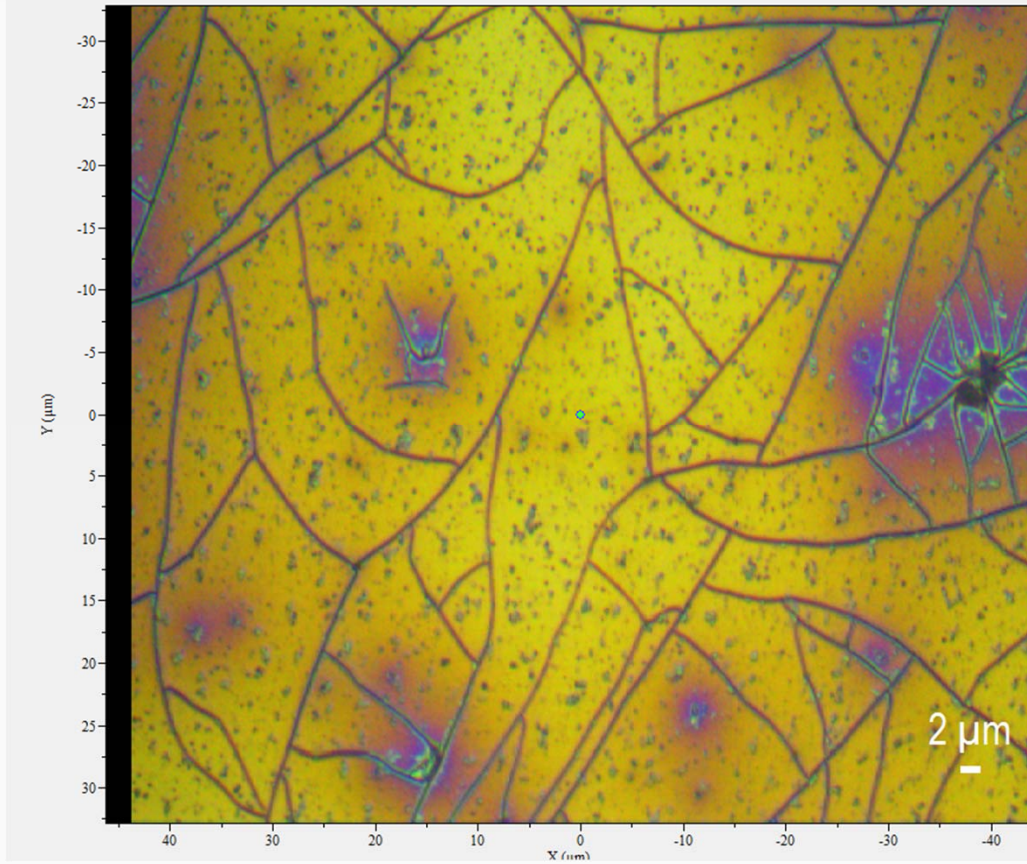
2, 4, 6, 8 dip trials in TiO_2 sol-gel

Raman spectroscopy imaging

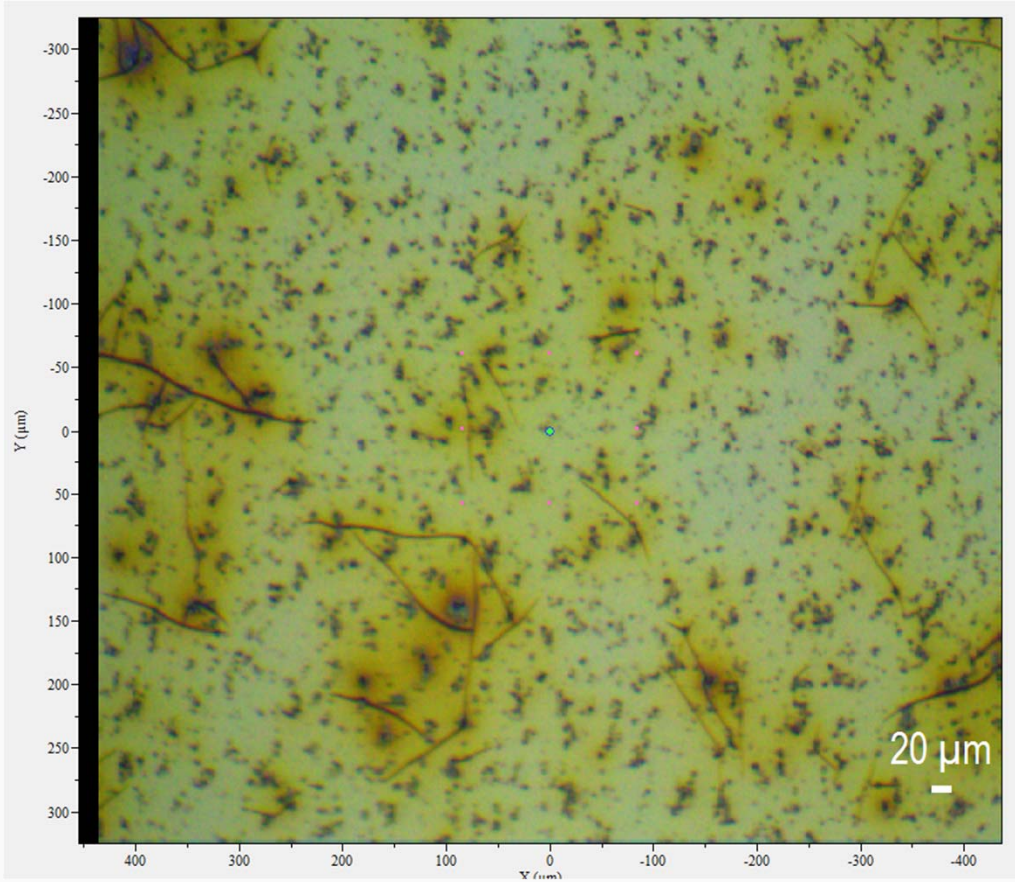
UV-Vis Methylene blue degradations



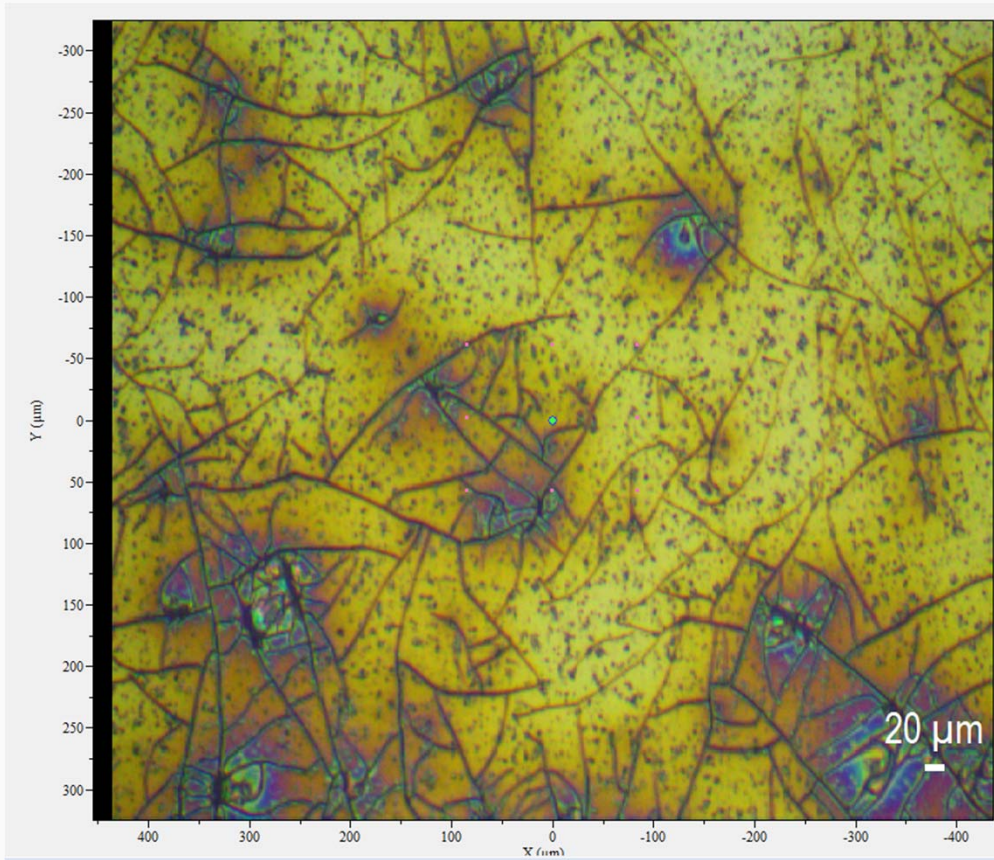
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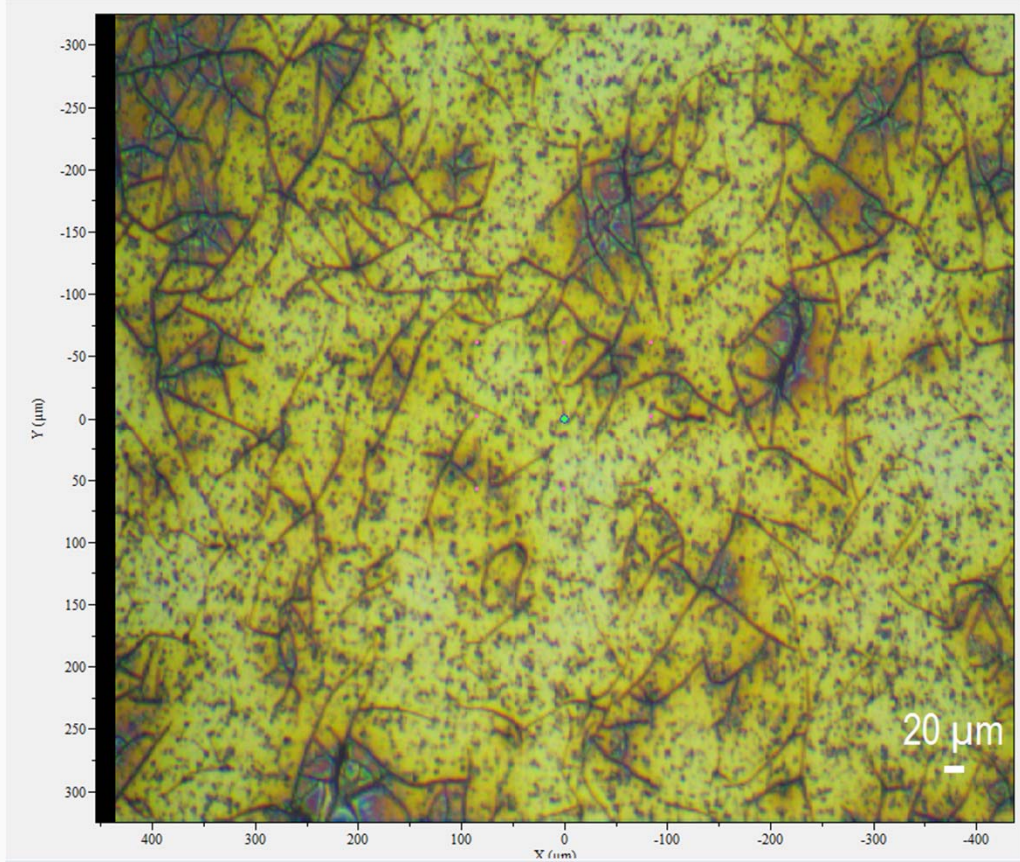
2 dips



4 dips

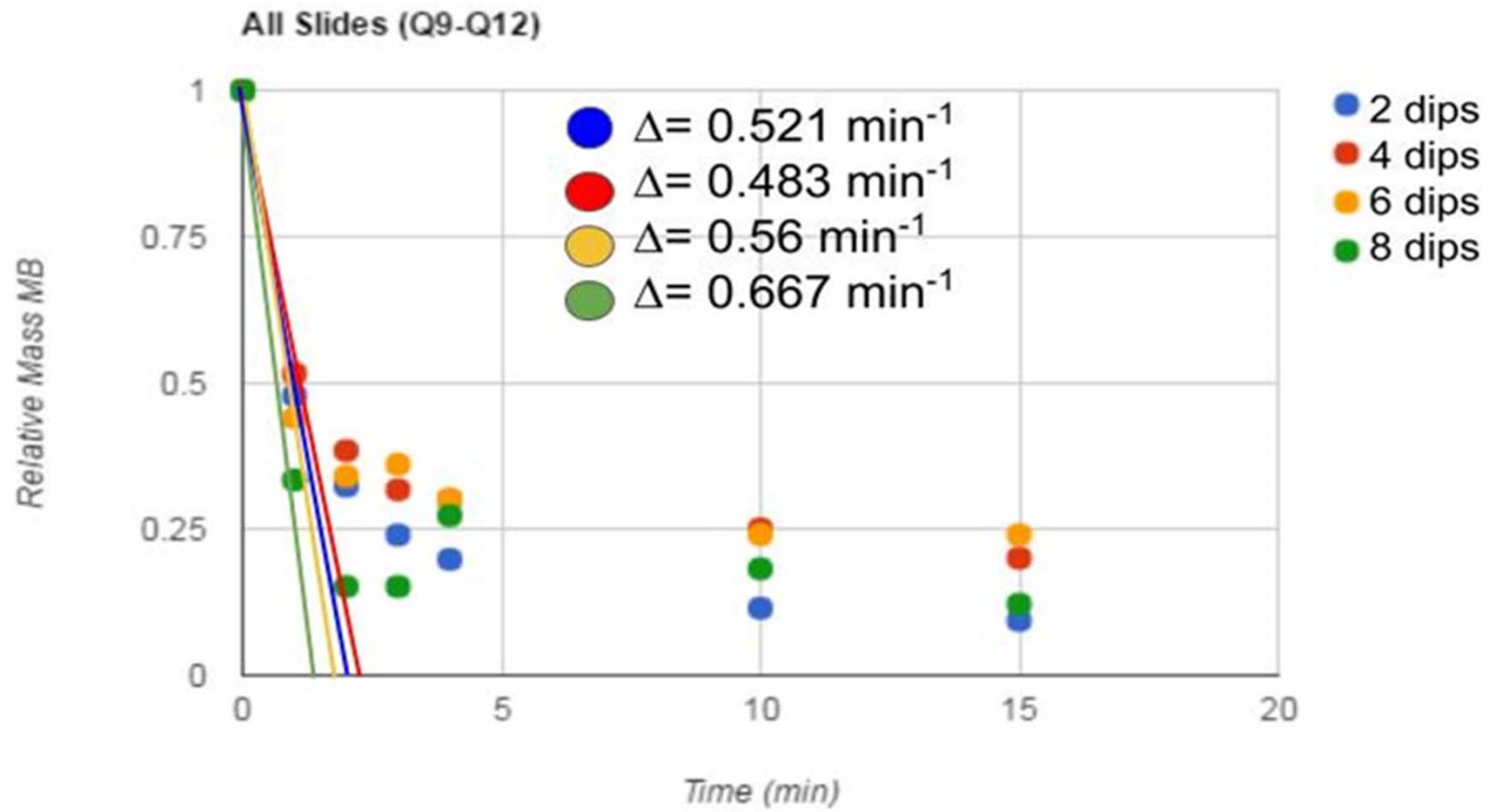


6 dips



8 dips

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Highly visible

All slides has some form of cracking and spotting

Drying the slides in a vacuum oven reduced, but did not eliminate, cracking

P25 particles are clearly visible on all slides

The 8 dip slide degraded the most pollutant during the first minute

G s r g p y w n s r

No certain thickness of coating prevents cracking on the surface of slide.

The thicker the sol-gel coating on the slide, the better its degradation ability on a pollutant.

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Dip coat quartz rods in the optimal sol-gel solution,
for a photocatalytic reactor

Run degradation trials within the reactor

Egors { pihkq irxw\$



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Ryan Catabay



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- [1] "Facts About Water & Sanitation." *Www.water.org*. Water.org, 2015. Web. 7 Aug. 2016.
- [2] "Facts and Figures." *2013*. UN Water, 2013. Web. 07 Aug. 2016.
- [3] Hashimoto, Kazuhito, Hiroshi Irie, and Akira Fujishima. "TiO₂ photocatalysis: a historical overview and future prospects." *Japanese journal of applied physics* 44.12R (2005): 8269.
- [4] Lee, Seul-Yi. "TiO₂ Photocatalyst for Water Treatment Applications." *TiO₂ Photocatalyst for Water Treatment Applications*. ScienceDirect, 16 July 2013. Web. 07 July 2016.
- [5] S. Fowler, R. Catabay, and J. Jiao. Optimization of Photocatalytic Films for Water Purification. [2015].